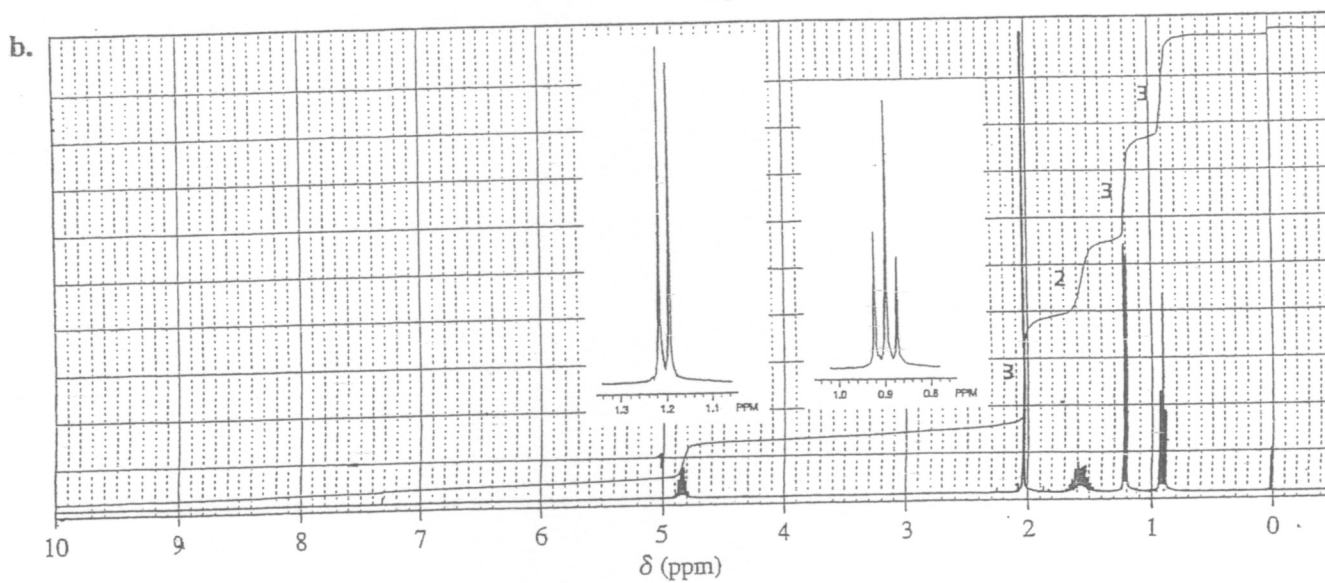
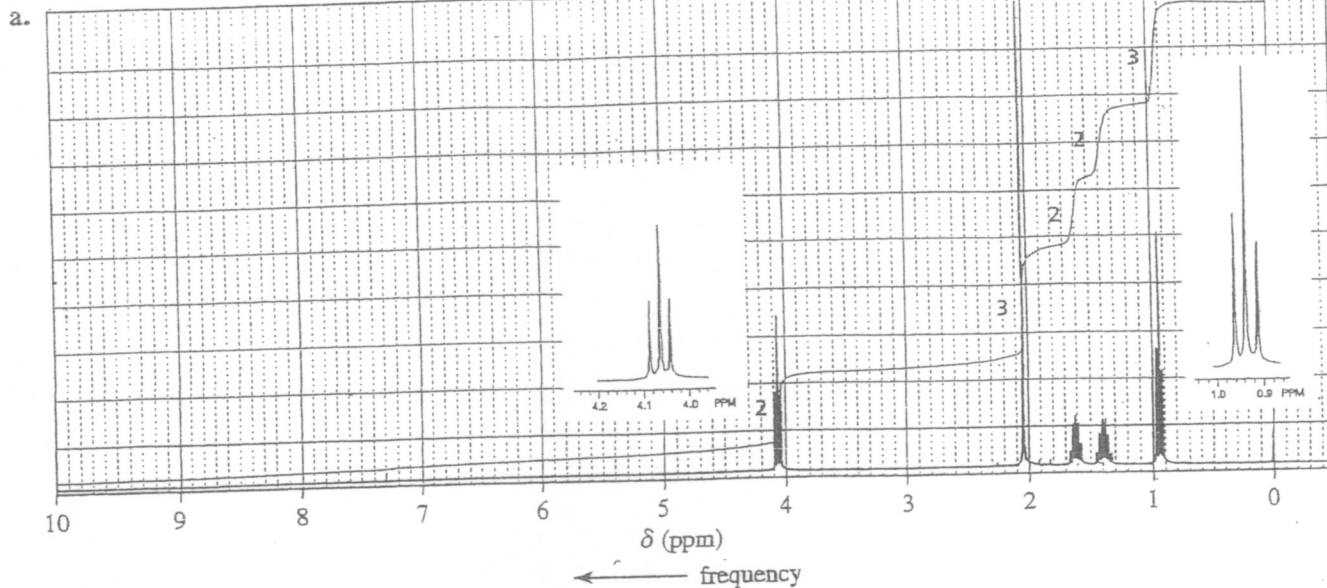
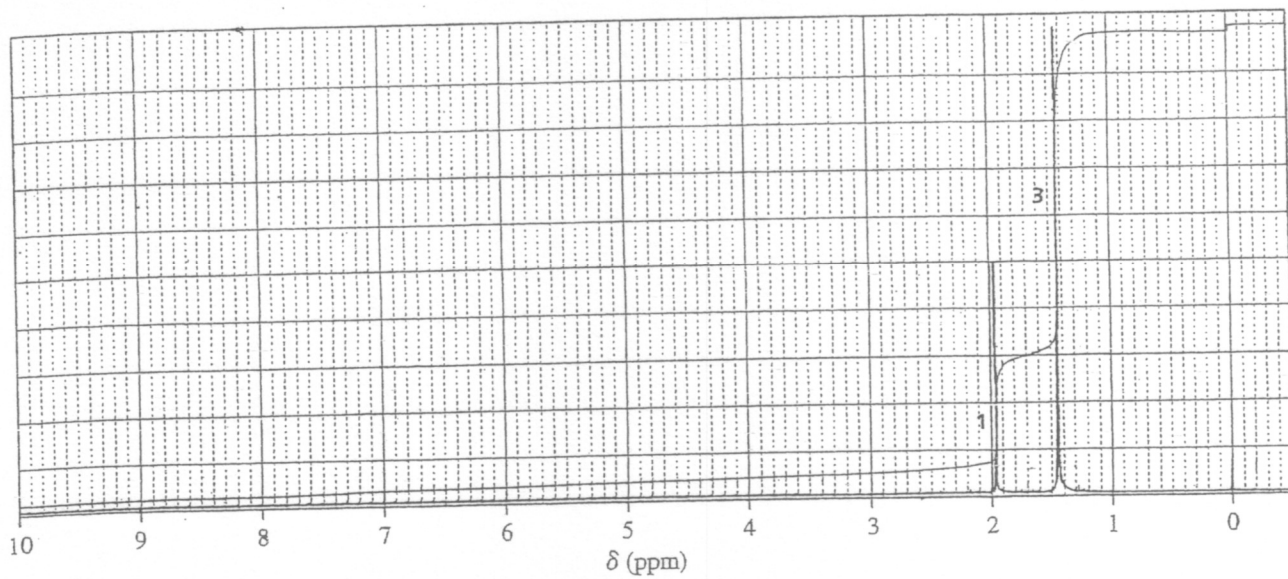
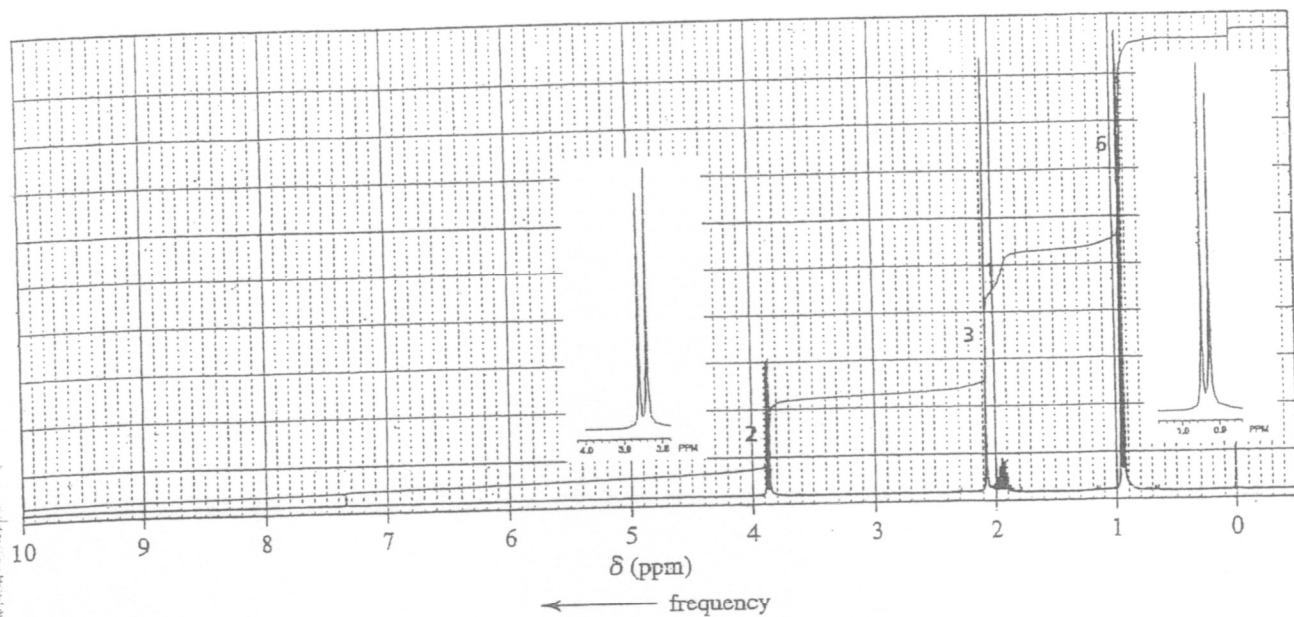


69. The following  $^1\text{H}$  NMR spectra are for four compounds each with molecular formula  $\text{C}_6\text{H}_{12}\text{O}_2$ . Identify the compounds.

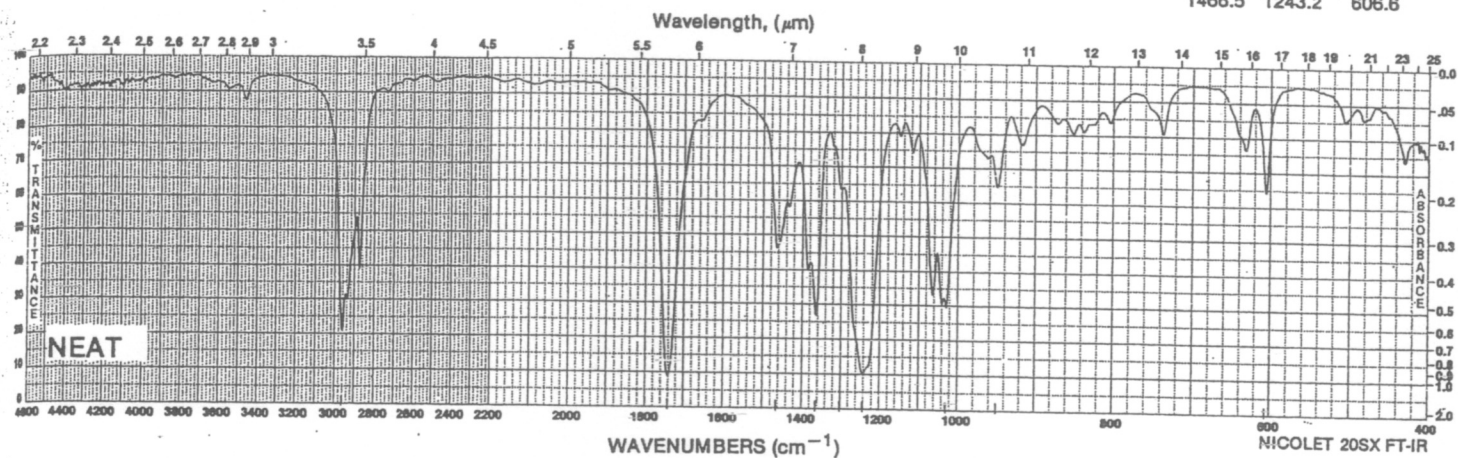


c.



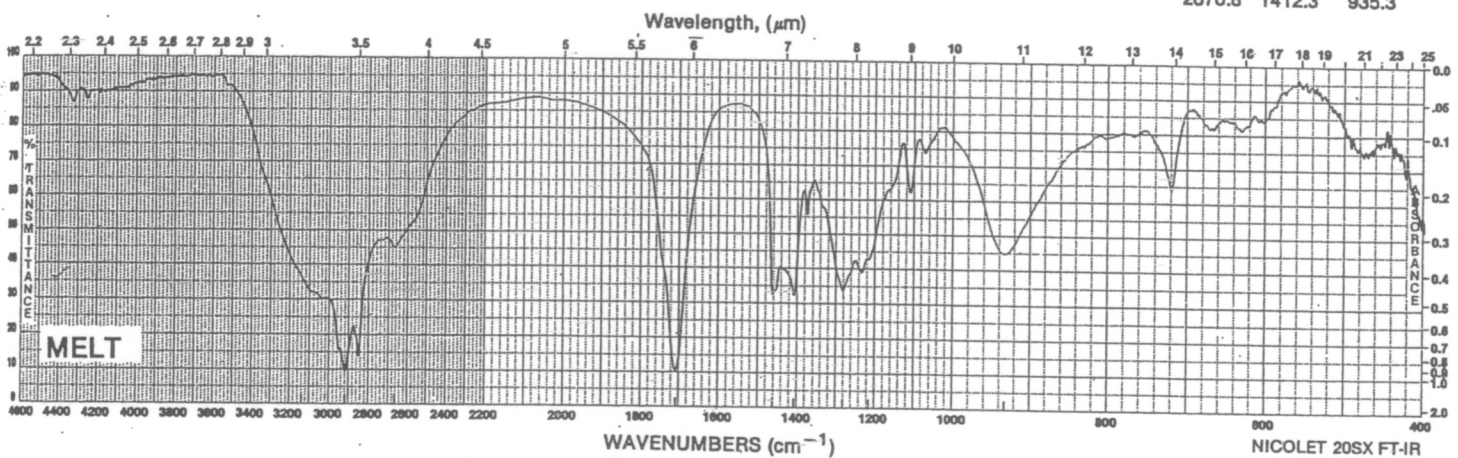
PROBLEM 3.4 SPECTRUM E

2962.1 1366.3 1031.3  
 1743.1 1303.6 950.8  
 1466.5 1243.2 606.6



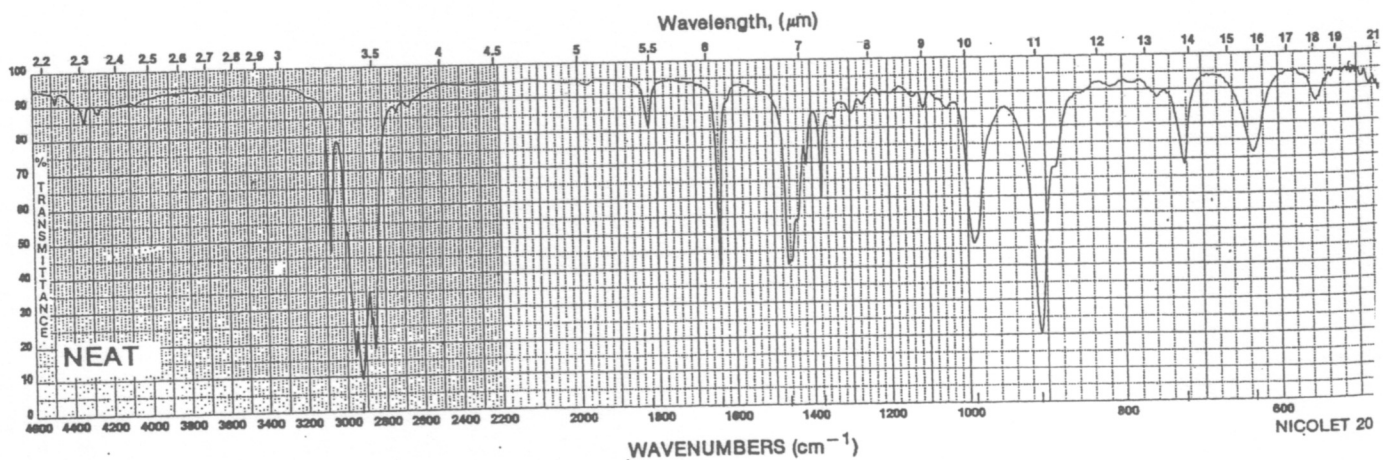
PROBLEM 3.4 SPECTRUM F

3148.2 1712.1 1285.0  
 2925.9 1466.3 1218.9  
 2870.8 1412.3 935.3



PROBLEM 3.3 SPECTRUM D

2927.1 1467.1 909  
 1822.5 1378.9 724  
 1641.7 993.3 634



**TABLE 13-3 Typical Values of Chemical Shifts**

Type of Proton	Approximate $\delta$	Type of Proton	Approximate $\delta$
alkane ( $-\text{CH}_3$ )	0.9	$\text{>C=C<}$ $\text{CH}_3$	1.7
alkane ( $-\text{CH}_2-$ )	1.3	Ph—H	7.2
alkane ( $-\text{CH}-$ )	1.4	Ph— $\text{CH}_3$	2.3
$\begin{array}{c} \text{O} \\ \parallel \\ -\text{C}-\text{CH}_3 \end{array}$	2.1	R—CHO	9–10
$-\text{C}\equiv\text{C}-\text{H}$	2.5	R—COOH	10–12
R— $\text{CH}_2$ —X (X = halogen, O)	3–4	R—OH	variable, about 2–5
$\begin{array}{c} \diagup \\ \text{C}=\text{C} \\ \diagdown \quad \text{H} \end{array}$	5–6	Ar—OH	variable, about 4–7
		R— $\text{NH}_2$	variable, about 1.5–4

Note: These values are approximate, as all chemical shifts are affected by neighboring substituents. The numbers given here assume that alkyl groups are the only other substituents present. A more complete table of chemical shifts appears in Appendix 1.

**TABLE 12-2 Summary of IR Stretching Frequencies**

Frequency ( $\text{cm}^{-1}$ )	Functional Group	Comments	
3300	alcohol amine, amide alkyne	$\text{O}-\text{H}$ $\text{N}-\text{H}$ $\equiv\text{C}-\text{H}$	always broad may be broad, sharp, or broad with spikes always sharp, usually strong
3000	alkane	$-\text{C}-\text{H}$	just below $3000 \text{ cm}^{-1}$
	alkene	$=\text{C}-\text{H}$	just above $3000 \text{ cm}^{-1}$
	acid	$\text{O}-\text{H}$	very broad
2200	alkyne nitrile	$-\text{C}\equiv\text{C}-$ $-\text{C}\equiv\text{N}$	just below $2200 \text{ cm}^{-1}$ just above $2200 \text{ cm}^{-1}$
1710 (very strong)	carbonyl	$\text{>C=O}$	ketones, aldehydes, acids esters higher, about $1735 \text{ cm}^{-1}$ conjugation lowers frequency amides lower, about $1650 \text{ cm}^{-1}$
1660	alkene	$\text{>C=C<}$	conjugation lowers frequency aromatic $\text{C}=\text{C}$ about $1600 \text{ cm}^{-1}$
	imine	$\text{>C=N<}$	stronger than $\text{C}=\text{C}$
	amide	$\text{>C=O}$	stronger than $\text{C}=\text{C}$ (see above)

Ethers, esters, and alcohols also show  $\text{C}-\text{O}$  stretching between  $1000$  and  $1200 \text{ cm}^{-1}$ .

ALDEHYDES 2700, 2800