

## **Chapter 12 Vocabulary**

Adhesion  
Amorphous solid  
Boiling point  
Closest packing  
Cohesion  
Condensation  
Coordination number  
Critical pressure  
Critical temperature  
Crystalline solid  
Deposition  
Dipole – dipole forces  
Dispersion forces  
Dynamic equilibrium  
Equilibrium vapor pressure  
Evaporation  
Hydrogen bond  
Induced dipole  
Intermolecular forces  
Intramolecular forces  
Ion – dipole forces  
Lattice point  
Melting point  
Molar heat of fusion  
Molar heat of sublimation  
Molar heat of vaporization  
Phase  
Phase changes  
Phase diagram  
Polarizability  
Sublimation  
Surface tension  
Triple point  
Unit cell  
Van der waals forces  
Vaporization  
Viscosity  
X-ray diffraction